MAR-15-2010 11:17 From:LYONDELLBASELL

16103592414

To: 915712738300

Page: 4/10

10/551,676 Response to Office Action of 12/15/2009.

RECEIVED
CENTRAL FAX CENTER

Listing of the Claims:

MAR 1 5 2010

This listing of claims will replace all prior versions and listings of claims in the application:

- 1. (Previously presented) A process for an olefin polymerization comprising (a) starting up the polymerization reaction in a gas-phase fluidized-bed reactor using a catalyst comprising a metallocene to produce a start-up polyolefin having a melt flow rate greater than 4.5 g/10 min; and (b) continuing the polymerization reaction and gradually decreasing the melt flow rate of the polyolefin to less than 4 g/10 min, wherein the melt flow rate is measured at 2.16 kg and 190°C in accordance with ISO 1133, and wherein said start-up phase of step (a) has a duration of 30 minutes to 30 hours.
- 2. (Previously presented) The process of claim 1, wherein the start-up phase of step (a) has a duration of 2 hours to 20 hours.
- 3. (Canceled).
- 4. (Previously presented) The process of claim 1, wherein the reaction temperature in step (a) is at least 1°C higher than the reaction temperature in step (b).
- 5. (Previously presented) The process of claim 4, wherein the reaction temperature in step (a) is 1.5 to 4°C higher than the reaction temperature in step (b).
- 6. (Previously presented) The process as claimed in claim 4, wherein the reaction temperature in step (b) is in a range bounded by an upper limit given by equation I

$$T_{RII} = 170 + \frac{6d'}{0.84 - d'} \tag{1}$$

and a lower limit given by equation II

$$T_{RN} = 173 + \frac{7.3d'}{0.837 - d'} \tag{II}$$

16103592414

To: 915712738300

Page: 5/10

10/551,676 Response to Office Action of 12/15/2009. Page 3

wherein,

T_{RH} is a maximum reaction temperature in °C

T_{RN} is a minimum reaction temperature in °C

d' is a value of a density of the polymer to be produced.

- 7. (Previously presented) The process of claim 1, wherein the melt flow rate is regulated by hydrogen concentration in the reactor.
- 8. (Previously presented) The process of claim 1, wherein the melt flow rate is regulated by a monomer partial pressure in the reactor.
- 9. (Previously presented) The process of claim 1, wherein the polyolefin is a homopolymer or copolymer of ethylene.
- 10. (Canceled).
- 11. (Currently amended) A process for an olefin polymerization comprising (a) starting up the polymerization reaction in a gas-phase fluidized-bed reactor using a catalyst comprising a metallocene to produce a start-up polyolefin having a melt flow rate greater than 4.5 g/10 min; and (b) continuing the polymerization reaction and gradually decreasing the melt flow rate of the polyolefin to less than 4 g/10 min, wherein the melt flow rate is measured at 2.16 kg and 190°C in accordance with ISO 1133, and wherein said start-up phase of step (a) has a duration of 30 minutes to 30 hours. The process as claimed in claim 1, wherein the metallocene is selected from bis(1-methyl-3-butylcyclopentadienyl)zirconium dichloride or bisindenylzirconium dichloride.
- 12. (Previously presented) The process of claim 1, wherein an alkylaluminoxane is used as an activating compound.
- 13. (New) The process of claim 11, wherein the start-up phase of step (a) has a duration of 2 hours to 20 hours.

MAR-15-2010	11:18	From: 1	YONDELL	BASELL

16103592414

To:915712738300

Page:6/10

- 14. (New) The process of claim 11, wherein the reaction temperature in step (a) is at least 1°C higher than the reaction temperature in step (b).
- 15. (New) The process of claim 14, wherein the reaction temperature in step (a) is 1.5 to 4°C higher than the reaction temperature in step (b).
- 16. (New) The process of claim 11, wherein the melt flow rate is regulated by hydrogen concentration in the reactor.
- 17. (New) The process of claim 11, wherein the melt flow rate is regulated by a monomer partial pressure in the reactor.
- 18. (New) The process of claim 11, wherein the polyolefin is a homopolymer or copolymer of ethylene.